

**Chemistry 101**  
**Solutions for Quiz # 1**  
**Tuesday, September 9th version**

1. (3 pts) Complete the following conversions:

(a) 187 lbs into kg

$$\frac{187 \text{ lbs}}{2.20 \text{ lb}} \times \frac{1 \text{ kg}}{2.20 \text{ lb}} = 85.0 \text{ kg}$$

(b) 35.00 ml into L

$$\frac{35.00 \text{ mL}}{1000 \text{ mL}} \times \frac{1 \text{ L}}{1000 \text{ mL}} = 0.03500 \text{ L}$$

*Note that the zeros at the end of a decimal number are significant, so they are retained in the final answer.*

(c) 0.150 mg into  $\mu\text{g}$

$$\frac{0.150 \text{ mg}}{1 \text{ mg}} \times \frac{1000 \mu\text{g}}{1 \text{ mg}} = 150. \mu\text{g}$$

*note that the zeros at the end of a decimal number are significant, so we have placed a "." after 150 to indicate it is significant and the answer has 3 s.f.*

2. (3 pts) For each measurement below, indicate the number of significant digits:

(a) 510.5 mg

*4, all non-zero digits are significant, zeroes between non-zero digits are significant.*

(b) 0.0605 L

*3, all non-zero digits are significant, zeroes at the beginning of a decimal number are not significant, zeroes between non-zero digits are significant.*

(c)  $2.0 \times 10^6$  dL

*2, all the digits in the coefficient of a number in scientific notation are significant.*

3. (2 pts) Give the answers to the following problems, rounding to the correct number of significant figures:

(a)  $1.03 + 10.5 + 10.881 = 22.4$

*addition - round to the smallest number of decimal places in the problem - in this case one.*

(b)  $1.52 \times 2.0 \times 5.2 = 16$

*multiplication - round to smallest # s.f. in problem - in this case two.*

4. (2 pts) A dialysis unit requires 75,000 mL of distilled water. How many gallons of water are needed?

$$\frac{75,000 \text{ mL}}{946 \text{ mL}} \times \frac{1 \text{ qt}}{946 \text{ mL}} \times \frac{1 \text{ gallon}}{4 \text{ qt}} = 20. \text{ gallons}$$

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**Possibly Helpful Conversions**

1 kg = 2.20 lb

454 g = 1 lb

1 ton = 907.2 kg

946 mL = 1 qt

0.946 L = 1 qt

1 L = 1.06 qt

1 gallon = 4 qts

1 mg = 1000  $\mu\text{g}$